

COMMENT

SUMMARY OF RECOMMENDATIONS OF AIPEA NOMENCLATURE COMMITTEE¹

Because of their small particle sizes and variable degrees of crystal perfection, it is not surprising that clay minerals proved extremely difficult to characterize adequately prior to the development of modern analytical techniques. Problems in characterization led quite naturally to problems in nomenclature, undoubtedly more so than for the macroscopic, more crystalline minerals. The popular adoption in the early 1950's of the X-ray powder diffractometer for clay studies helped to solve some of the problems of identification. Improvements in electron microscopic, electron diffraction and oblique-texture electron diffraction, infrared, and differential thermal equipment, and the development of nuclear and isotope technology, high-speed electronic computers, Mössbauer spectrometers, and most recently of electron microprobes and scanning electron microscopes all have aided in the accumulation of factual information on clays. This, in turn, should facilitate eventual agreement on the nomenclature of clays.

Probably the earliest attempt by clay scientists to reach agreement on nomenclature and classification on an international basis was at the International Soil Congress held in Amsterdam in 1950 (Brindley *et al.*, 1951). Since that time national nomenclature committees have been established in many countries. Recommendations from these national groups have been considered every three years at the International Clay Conferences, first by the Nomenclature Sub-Committee of CIPEA (Comité International pour l'Etude des Argiles) and since 1966 by the Nomenclature Committee of AIPEA (Association Internationale pour l'Etude des Argiles). These international committees in turn have worked closely with the Commission on New Minerals and Mineral Names of the International Mineralogical Association (I.M.A.).

This summary of the recommendations made to date by the international nomenclature committees has been prepared to achieve wider dissemination of the decisions reached and to aid clay scientists in the correct usage of clay nomenclature. Some of the material in the present summary has been taken from an earlier summary by Bailey *et al.* (1971a).

CLASSIFICATION

Agreement was reached early in the international discussions that a sound nomenclature is necessarily based on a satisfactory classification scheme. For this reason, the earliest and most extensive efforts of the several national nomenclature committees were expended on classification schemes. Existing schemes were collated and discussed (see Brown, 1955; Mackenzie, 1959; and Pedro, 1967, for examples), symposia were held at national meetings, and polls were taken of clay scientists in 32 countries as to their preferences. Armed with these data, the international representatives were able to agree upon most features of a broadly based scheme for the phyllosilicates as a whole (Mackenzie, 1965a, 1965b; Brindley, 1967).

Table 1 gives the classification scheme in its present form. The phyllosilicates are divided into groups, each containing dioctahedral and trioctahedral subgroups. Each subgroup in

turn is divided into mineral species. This subdivision corresponds to successive stages of refinement in the identification process. It is anticipated that the precise definitions of the groups and subgroups and their names will evolve and change with time. This table differs from previously published versions in two respects. *Smectite* has now been accepted as the group name for clay minerals with layer charge between 0.2 and 0.6 per formula unit. This decision, made at the 1975 Mexico City meeting (Brindley and Pedro, 1976), was based on increased usage worldwide of this name rather than of the alternate dual name of *montmorillonite-saponite* for the group. Dual names still exist for the *kaolinite-serpentine* and *pyrophyllite-talc* groups. Suggested names of *kandite* and *septechlorite* for the kaolin and serpentine minerals, respectively, have not been approved by the AIPEA Committee, and should not be used. The second change is to treat chlorite as consisting of a 2:1 layer plus an interlayer hydroxide sheet, rather than as a 2:1:1 or 2:2 layer type. This emphasizes the similarity of chlorite to other clay minerals containing interlayer material (Brindley and Pedro, 1972).

DEFINITION OF PHYLLOSILICATE

Table 1 assumes a specific definition of a phyllosilicate (or layer silicate). This definition was discussed most recently at the AIPEA Nomenclature Committee meeting held in Madrid in 1972, at which a 1969 definition was modified. The present definition (Brindley and Pedro, 1972) states "Clay minerals belong to the family of phyllosilicates and contain continuous two-dimensional tetrahedral sheets of composition T_2O_5 ($T = Si, Al, Be, \dots$) with tetrahedra linked by sharing three corners of each, and with the fourth corner pointing in any direction. The tetrahedral sheets are linked in the unit structure to octahedral sheets, or to groups of coordinated cations, or individual cations." The present definition is based on the nature of the silicate parts of the structure, and does not include previous requirements of weaker interlayer bonding or of certain resultant physical properties. Thus, it does not require a category of "pseudo-layer silicates" for minerals, such as palygorskite and sepiolite, that do not possess marked basal cleavages. The criterion of a continuous tetrahedral sheet does exclude "quasi-layer silicates," such as astrophyllite, lamprophyllite, bafertisite, and haradaite, in which 5-fold or 6-fold coordinated groups interrupt the continuity of the tetrahedral net.

STANDARDIZATION OF STRUCTURAL TERMS

At the 1975 Mexico City meeting the AIPEA Nomenclature Committee noted that "lattice" and "structure" continue to be misused by authors and speakers. A "lattice" is not synonymous with "structure," but is a uniform distribution of points in space (e.g., the 14 Bravais lattices). The terms "layer lattice" and "Schichtgitter" are incorrect and should not be used. Layer structure, layer silicate, and phyllosilicate are acceptable terms (Brindley, 1967; Brindley and Pedro, 1976).

In 1972 the Committee agreed upon usage of the terms "plane," "sheet," "layer," "unit structure," and their equivalents in other languages (Brindley and Pedro, 1972). Recommended usage is as a single *plane* of atoms, a tetrahedral or octahedral *sheet*, and a 1:1 or 2:1 *layer*. Thus, plane, sheet, and layer refer to increasingly thicker arrangements. A sheet is a combination of planes, and a layer is a combination of sheets. In addition, layers may be separated from one another

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Table 1. Classification scheme for phyllosilicates related to clay minerals.

Layer type	Group (x = charge per formula unit)	Subgroup	Species ¹
1:1	Kaolinite-serpentine $x \sim 0$	Kaolinite Serpentine	Kaolinite, dickite, halloysite Chrysotile, lizardite, amesite
	Pyrophyllite-talc $x \sim 0$	Pyrophyllite Talc	Pyrophyllite Talc
2:1	Smectite $x \sim 0.2-0.6$	Di octahedral smectite Tri octahedral smectite	Montmorillonite, beidellite Saponite, hectorite, saunconite
	Vermiculite $x \sim 0.6-0.9$	Di octahedral vermiculite Tri octahedral vermiculite	Di octahedral vermiculite Tri octahedral vermiculite
	Mica ² $x \sim 1$	Di octahedral mica Tri octahedral mica	Muscovite, paragonite Phlogopite, biotite, lepidolite
	Brittle mica $x \sim 2$	Di octahedral brittle mica Tri octahedral brittle mica	Margarite Clintonite, anandite
	Chlorite x variable	Di octahedral chlorite Di, tri octahedral chlorite Tri octahedral chlorite	Donbassite Cookeite, sudoite Clinochlore, chamosite, nimitite

¹ Only a few examples are given.

² The status of *illite* (or *hydromica*), *sericite*, etc. must be left open at present, because it is not clear whether or at what level they would enter the table: many materials so designated may be interstratified.

by various *interlayer* materials, including cations, hydrated cations, organic molecules, and hydroxide octahedral groups and sheets. The total assembly of a layer plus interlayer material is referred to as a *unit structure*. Table 2 lists the equivalent terms in other languages, as modified at the 1978 Oxford meeting.

The terms "talc layer" and "brucite sheet" are not suitable for describing the component parts of the chlorite structure, because the minerals talc and brucite permit very little substitution of Mg by Al, which is an essential feature of trioctahedral chlorites. It is recommended that *2:1 layer* be used in place of "talc layer" and *hydroxide sheet* or *interlayer sheet* in place of "brucite sheet" (Brindley and Pedro, 1972). It is permissible to write *brucite-like* or *brucitic* or *gibbsite-like* or *gibbsitic* if one wishes to specify the trioctahedral or dioctahedral nature of the interlayer.

Attention is drawn also to the report of the I.M.A.-I.U.Cr. Joint Committee on Nomenclature (Bailey, 1977), in which the following recommendations will be of special interest to clay scientists. These recommendations have been approved by the AIPEA Nomenclature Committee.

1. *Polytypism* is defined as "the phenomenon of the existence of an element or compound in two or more layer-like

crystal structures that differ in layer stacking sequences. The layers need not be crystallographically identical, but should be similar. Polytypism differs from polymorphism (in the present and strict definition of the latter term) in permitting small differences in chemical composition between structures, not to exceed 0.25 atoms per formula unit of any constituent element. Layer structures that differ from one another by more than this amount are to be called *polytypoids* rather than *polytypes*."

2. "In general, polytypes should not receive individual mineral names. Instead, a set of related polytypes should be designated by a single name followed by a structural symbol suffix that defines the layer stacking differences." A recommended system of structural symbols is described in the report.

3. "Polytype mineral names already in existence that have international acceptance and serve a useful function need not be discarded. Decision on retention of individual names should be the responsibility of the I.M.A. Commission on New Minerals and Mineral Names."

4. "It is recommended that X , Y , Z , or $[100]$, $[010]$, $[001]$ be used for directions of crystallographic axes and a , b , c for the repeat distances along these axes."

Table 2. Structural terms of reference and their equivalents in different languages.

English	French	German	Russian	Spanish	Italian
plane	plan	Ebene	плоскость	plano	plano
sheet	couche	Schicht	сетка	capa	strato
layer	feuille	Schichtpaket	слой	estrato o paquete (de capas)	pacchetto
interlayer	espace interfoliaire	Zwischenschicht	межслоево промежуток (межслой)	material interlaminar	interstrato
unit structure	unité structurale	Struktur Einheit	пакет	unidad estructural	unita strutturale

INTERSTRATIFICATIONS AND NON-CRYSTALLINE MATERIALS

No general agreement has been reached yet as to preferred terminology for interstratified minerals, except that the material should be characterized fully as to degree of regularity or irregularity of the interstratification and that it should be described in terms of the nature and ratios of the component layers. The best descriptive terms for those layers are still in question. At the 1972 Madrid meeting the Committee recommended that specific names not be given to poorly defined materials, such as irregularly interstratified systems, imperfect structures (e.g., deweylite and aquacryptite), or to non-crystalline constituents. Special names can be given to regularly interstratified minerals, subject to acceptance by the AIPEA Nomenclature Committee and the I.M.A. Commission on New Minerals and Mineral Names (Brindley and Pedro, 1972). Names already in the literature at that time were *rectorite* for a regular 1:1 interstratification of dioctahedral paragonite/smectite (Brown and Weir, 1963), *corrensite* for a regular 1:1 interstratification of trioctahedral chlorite/"swelling chlorite" (Lippmann, 1954), *tosudite* for a regular 1:1 interstratification of dioctahedral chlorite/smectite (Frank-Kamenetskii *et al.*, 1963; Shimoda, 1969), and *aliyttite* for a regular 1:1 interstratification of trioctahedral talc/saponite (Veniale and van der Marel, 1969). The I.M.A. Commission on New Minerals and Mineral Names has disapproved the name *sangarite* for a regular 1:1 interstratification of trioctahedral chlorite/vermiculite (Drits and Kossovskaya, 1963), and has approved the name *tarasovite* for a regular 3:1 interstratification of dioctahedral mica/smectite (Lazarenko and Korolev, 1970). The AIPEA Nomenclature Committee has taken no action as yet on specific names for regular interstratifications.

The CIPEA Nomenclature Sub-Committee at its Jerusalem meeting (Brindley, 1967) agreed unanimously that the term "non-crystalline" is preferable to the commonly used term "amorphous." It was recommended strongly that specific names not be given to newly discovered non-crystalline minerals, but that they be described so far as possible in terms of their chemical composition. Names may be chosen later if it becomes apparent that particular ranges of chemical composition exist for these minerals.

SPECIFIC PHYLLOSILICATE NAMES

Dioctahedral chlorite

The Committee has recommended (Brindley and Pedro, 1970) that the chlorite group be subdivided into the three subgroups dioctahedral chlorite, di, trioctahedral chlorite, and trioctahedral chlorite (Table 1). *Dioctahedral chlorite* is dioctahedral in both the 2:1 layer and the interlayer hydroxide sheet. An example is *donbassite* (Lazarenko, 1940). *Trioctahedral chlorite* is trioctahedral in both octahedral sheets. A *di, trioctahedral chlorite* is dioctahedral in the 2:1 layer but trioctahedral in the interlayer sheet. *Cookeite* and *sudoite* are examples of di, trioctahedral chlorites, with cookeite being Li-rich and sudoite Li-poor. No examples are known as yet of chlorites with trioctahedral 2:1 layers but dioctahedral interlayers.

Trioctahedral chlorite

At the 1978 Oxford meeting the AIPEA Nomenclature Committee adopted the suggestion of Bayliss (1975) for simplification of chlorite nomenclature. Trioctahedral chlorites should be named according to the dominant divalent octahedral cation present. Recommended species names are *clinocllore* for Mg-dominant [end member = $(Mg_5Al)(Si_3Al)O_{10}(OH)_8$], *chamosite* for Fe²⁺-dominant [end member =

$(Fe^{2+}_5Al)(Si_3Al)O_{10}(OH)_8$], *nimitte* for Ni-dominant [end member = $(Ni_5Al)(Si_3Al)O_{10}(OH)_8$], and *pennantite* for Mn²⁺-dominant [end member = $(Mn^{2+}_5Al)(Si_3Al)O_{10}(OH)_8$]. All other species and varietal names should be discarded because arbitrary subdivisions according to octahedral and tetrahedral compositions have been shown to have little or no structural significance. Tetrahedral compositions and trivalent octahedral cations are not considered in the recommended species names, nor is the distribution of octahedral cations between the 2:1 layer and the interlayer. Adjectival modifiers, such as those of Schaller (1930), may be used to indicate either important octahedral cations other than the dominant cation or unusual tetrahedral compositions. Bayliss (1975) gives modifiers appropriate for many of the chlorite species listed in other nomenclature systems.

Imogolite

The Committee at its 1969 Tokyo meeting (Brindley and Pedro, 1970) approved the name *imogolite* for a hydrous aluminosilicate having a fine threadlike morphology and the diffraction characteristics described by Wada and Yoshinaga (1969) and by others.

Halloysite

The 1975 AIPEA Nomenclature Committee reviewed the several terminologies in use for the less hydrous and the more hydrous forms of halloysite. The terms *halloysite*(7Å) and *halloysite*(10Å) were recommended for general usage as being least ambiguous (Brindley and Pedro, 1976). Therefore, the term *endellite* should not be used.

Celadonite

The 1978 AIPEA Nomenclature Committee has defined *celadonite* as a dioctahedral mica of ideal composition $KMgFe^{3+}Si_4O_{10}(OH)_2$ but allowing a tetrahedral Al (or Fe³⁺) range of 0.0 to about 0.2 atoms per formula unit. Substantial octahedral variations from this formula can be described by adjectival modifiers, such as aluminian celadonite or ferroan celadonite. Further characteristics of celadonite are $d(060) < 1.510$ Å and sharp infrared spectra, as described by Buckley *et al.* (1978). There is an area of potential overlap of celadonite and glauconite analyses between about $Al^{IV} = 0.17$ to 0.20 atoms. For compositions near this boundary and for cases where analytical errors or impurities are suspected, application of the other identification criteria are especially important.

Glauconite

Buckley *et al.* (1978) have shown that with careful purification and modern analytical techniques there is little or no overlap between celadonite and glauconite compositions and that the two minerals can be differentiated also by $d(060)$ values and infrared spectra. The 1978 AIPEA Nomenclature Committee defined glauconite as an Fe-rich dioctahedral mica with tetrahedral Al (or Fe³⁺) usually greater than 0.2 atoms per formula unit and octahedral R³⁺ correspondingly greater than 1.2 atoms. A generalized formula is $K(R^{3+}_{1.33}R^{2+}_{0.67})(Si_{3.67}Al_{0.33})O_{10}(OH)_2$ with $Fe^{3+} \gg Al$ and $Mg > Fe^{2+}$ (unless altered). Further characteristics of glauconite are $d(060) > 1.510$ Å and (usually) broader infrared spectra than celadonite, as described by Buckley *et al.* (1978). The species glauconite is single-phase and ideally is non-interstratified. Mixtures containing an iron-rich mica as a major component can be called *glauconitic*. Specimens with expandable layers can be described as randomly interstratified glauconite/smectite. Mode of origin is not a criterion, and a

green fecal pellet in a marine sediment that meets the definition for celadonite should be called celadonite.

Miscellaneous

Attention is drawn here to recommendations made by other nomenclature committees, although not specifically considered by the AIPEA Nomenclature Committee.

The name *berthierine* has priority for the Fe-rich 1:1 type layer silicate having appreciable tetrahedral Al and commonly found in ironstones and iron formations. *Brindleyite* is the Ni-analogue of berthierine. The name *chamosite* has priority for a 2:1 chlorite of composition similar to berthierine (Orcel *et al.*, 1949).

The name *clintonite* has priority over other species names (xanthophyllite, seybertite, brandisite, valuevite) for the Li-poor, Ba-poor trioctahedral brittle mica. All of these are so similar in crystallography, chemical composition, and mode of origin that only a single species name is justified (Forman *et al.*, 1967). *Bityite* (Li, Be-rich), *anandite* (Ba, Fe-rich), and *kinoshitalite* (Ba, Mg-rich) appear to be other valid trioctahedral brittle mica species (Schaller *et al.*, 1967; Pattiaratchi *et al.*, 1967; Yoshii *et al.*, 1973). *Ephesite*, described originally as a Li-Na brittle mica (Schaller *et al.*, 1967), is described better as a true mica with a layer charge per formula unit of unity.

The name *palygorskite* has priority over *attapulgit* for the mineral with ribbonlike structure in which the ribbons have a width of two pyroxene-like chains (Bailey *et al.*, 1971b).

The name *anauxite* has been discredited. It is a mixture of components, of which the kaolin component is true kaolinite (Langston and Pask, 1968; Allen *et al.*, 1969; Bailey and Langston, 1969).

Medmontite is a mixture of chrysocolla and mica, and the name should be discarded (Chukhrov *et al.*, 1968, 1969; Fleischer, 1969a).

Nimite is the preferred term for the trioctahedral chlorite with Ni-dominant (Hiemstra and de Waal, 1968a). Specimens previously termed *schuchardite* have Ni < Mg (Fleischer, 1969b), and should be called nickeloan clinochlore. Brindley and De Souza (1975) also have shown that some "schuchardites" are transitional between chlorite and vermiculite.

Caryophilite is the preferred term for a 1:1 layer type mineral that is the Mn²⁺-analogue of greenalite. The name *bementite*, sometimes used for the former mineral, has priority for a Mn-rich mineral that belongs to the frieldite group of minerals and is not a layer silicate (Kato, 1963).

The name *rectorite* has priority over *allevardite* for a regular 1:1 interstratification of paragonite/smectite (Brown and Weir, 1963).

Sungulite and *kolskite* are mixtures of lizardite and sepiolite, and the names should be discarded (Ivanova *et al.*, 1973).

Alushtite is a mixture of dickite and hydrous mica, and the name should be discarded (Logvinenko and Frank-Kamenetskii, 1955). Some specimens that have been called alushtite have been identified later as tosudite (Frank-Kamenetskii *et al.*, 1963).

Deweylite is a mixture in variable proportions of a disordered form of talc (kerolite) and a disordered form of serpentine. Both components have excess water, probably associated with unbalanced surface bonds. The name is useful only as a field term (Bish and Brindley, 1978).

Kerolite is a varietal name for a mineral close to talc in composition and structure but with highly random layer stacking and an enlarged basal spacing of about 9.6 Å due to misfitting layers. R²⁺₃(Si₂O₅)₂(OH)₂·nH₂O with n ~ 0.8–1.2 (Brindley *et al.*, 1977).

Pimelite is a Ni-analogue of kerolite with Ni > Mg (Maksimović, 1966; Brindley *et al.*, 1979).

Nepouite is a Ni-analogue of lizardite (Glasser, 1907; Maksimović, 1973; Brindley and Wan, 1975).

New names for layer silicate minerals approved recently by the I.M.A. Commission on New Minerals and Mineral Names are listed below:

hendricksite, a trioctahedral Zn-rich mica (Fron del and Ito, 1966; Fron del and Einaudi, 1968)

willemseite, a Ni-analogue of talc (Hiemstra and de Waal, 1968b)

pecoraite, a Ni-analogue of clinochrysotile (Faust *et al.*, 1969)

Mn-sepiolite, *Mn-palygorskite*, *Mn-ferrisepiolite*, *Mn-ferropalygorskite* (Semenov, 1969)

chernykhite, a dioctahedral V, Ba, Na-rich mica (Ankinovich *et al.*, 1972)

kellyite, a Mn²⁺-analogue of amesite (Peacor *et al.*, 1974)

swinefordite, a Li, Al, Mg-rich smectite intermediate between dioctahedral and trioctahedral (Tien *et al.*, 1975)

baumite, a Mn, Fe, Zn-rich serpentine (Fron del and Ito, 1975)

masutomilite, a Mn²⁺-analogue of zinnwaldite (Harada *et al.*, 1976).

yofortierite, a Mn²⁺-analogue of palygorskite (Perrault *et al.*, 1975).

falcondoite, a Ni-analogue of sepiolite with Ni > Mg (Springer, 1976).

ferripyrophyllite, a Fe³⁺-analogue of pyrophyllite (Chukhrov *et al.*, 1979).

Several layer silicates incorporating interlayer metallic elements have been recognized recently.

Chapmanite and *bismutoferrite* have 1:1 layers with Si in the tetrahedral sheet and Fe³⁺ in the octahedral sheet. The surface hydroxyl groups of the octahedral sheet are replaced by oxygens, and Sb and Bi (in chapmanite and bismutoferrite, respectively) are in the interlayer space (Zhoukhlistov *et al.*, 1974; Zhoukhlistov and Zvyagin, 1977).

Surite is a smectite having a defect, cerussite-like lead carbonate interlayer (Hayase *et al.*, 1978).

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² The correct adjectival modifier is nickeloan.

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