

Crystal structure of ethynodiol diacetate, C₂₄H₃₂O₄James A. Kaduk^{1,2} , Anja Dosen³  and Tom N. Blanton³ ¹Department of Chemistry, Illinois Institute of Technology, 3101 South Dearborn Street, Chicago, IL 60616, USA²Department of Physics, North Central College, 131 South Loomis Street, Naperville, IL 60540, USA³International Centre for Diffraction Data (ICDD), 12 Campus Boulevard, Newtown Square, PA 19073-3273, USA

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Abstract: The crystal structure of ethynodiol diacetate has been solved and refined using synchrotron X-ray powder diffraction data and optimized using density functional theory techniques. Ethynodiol diacetate crystallizes in space group $P2_1$ (#4) with $a = 17.4055(12)$, $b = 7.25631(17)$, $c = 19.6008(14)$ Å, $\beta = 116.2471(23)^\circ$, $V = 2,220.33(13)$ Å³, and $Z = 4$ at 298 K. The crystal structure consists of alternating layers of the two independent molecules parallel to the (101) plane. The molecules do not interact strongly with each other, as reflected by the low density of 1.150 g/cm³. The powder pattern has been submitted to the International Centre for Diffraction Data (ICDD) for inclusion in the Powder Diffraction File™ (PDF®).

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Key words: ethynodiol diacetate, Ovulen, crystal structure, Rietveld refinement, density functional theory

I. INTRODUCTION

Ethynodiol diacetate (also known and spelled as etynodiol diacetate, sold under the brand names Ovulen, Femulen, and Continuin among others) is a progestin used in birth control pills and functions in preventing ovulation. The systematic name (CAS Registry Number 297-76-7; C₂₄H₃₂O₄, 384.26 g/mol) is [(3*S*,8*R*,9*S*,10*R*,13*S*,14*S*,17*R*)-17-acetyloxy-17-ethynyl-13-methyl-2,3,6,7,8,9,10,11,12,14,15,16-dodecahydro-1H-cyclopenta[*a*]phenanthren-3-yl] acetate. A two-dimensional molecular diagram of ethynodiol diacetate is shown in Figure 1.

Ethynodiol diacetate was described as a potent oral inhibitor of ovulation by Pincus et al. (1962). The crystal structure of a biotransformation product of ethynodiol diacetate was reported by Zafar et al. (2012). Crystal structures of many related compounds (particularly with a carbonyl group on the 6-membered ring of the steroid system at what we name C23) have been reported, but we are unaware of any published crystal structures or powder diffraction data for ethynodiol diacetate.

This work was carried out as part of a project (Kaduk et al., 2014) to determine the crystal structures of large-volume commercial pharmaceuticals and include high-quality powder diffraction data for them in the Powder Diffraction File (Kabekkodu et al., 2024).

II. EXPERIMENTAL

Ethynodiol diacetate was a commercial reagent, purchased from TargetMol (Batch #T0996), and was used as

received. The white powder was packed into a 0.5-mm-diameter Kapton capillary and rotated during the measurement at ~2 Hz. The powder pattern was measured at 298(1) K at the Wiggler Low Energy Beamline (Leontowich et al., 2021) of the Brockhouse X-ray Diffraction and Scattering Sector of the Canadian Light Source using a wavelength of 0.819826(2) Å (15.1 keV) from 1.6 to 75.0° 2θ with a step size of 0.0025° and an overall collection time of 3 minutes. The high-resolution powder diffraction data were collected using eight Dectris Mythen2 X series 1 K linear strip detectors. NIST SRM 660b LaB₆ was used to calibrate the instrument and refine the monochromatic wavelength used in the experiment.

The pattern was indexed using N-TREOR as incorporated into EXPO2014 (Altomare et al., 2013) on a primitive monoclinic unit cell with $a = 17.41661$, $b = 7.25165$, $c = 19.61664$ Å, $\beta = 116.233^\circ$, $V = 2,222.4$ Å³, and $Z = 4$. The suggested space group was $P2_1$, which was confirmed by the successful solution and refinement of the structure. A reduced cell search of the Cambridge Structural Database (Groom et al., 2016) yielded 4 hits, but no structures of ethynodiol diacetate or its derivatives.

An ethynodiol diacetate molecule was downloaded from PubChem (Kim et al., 2023) as Conformer3D_COMPOUND_CID_9270.sdf. It was converted to a *.mol2 file using Mercury (Macrae et al., 2020) and into a Fenske–Hall Z-matrix using OpenBabel (O’Boyle et al., 2011). The crystal structure was solved using parallel tempering as implemented in FOX (Favre-Nicolin and Cerný, 2002), using $(\sin\theta/\lambda)_{\max} = 0.32$ Å⁻¹ and two ethynodiol diacetate molecules as fragments.

Rietveld refinement was carried out with General Structure Analysis System, II (GSAS-II) (Toby and Von Dreele, 2013). Only the 2.0–37.0° portion of the pattern was included

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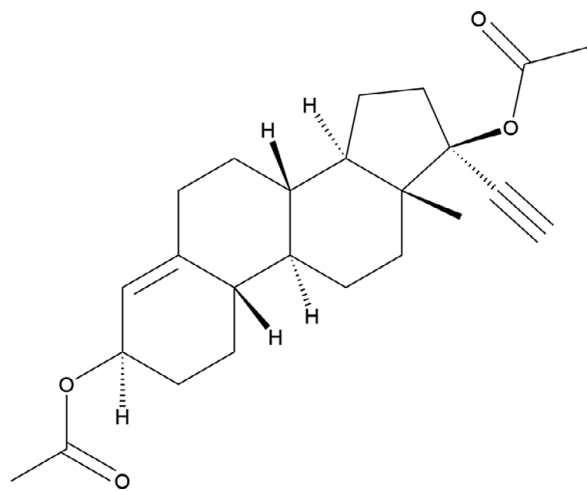


Figure 1. The two-dimensional structure of ethynodiol diacetate.

in the refinements ($d_{\min} = 1.292 \text{ \AA}$). The y-coordinate of O1 was fixed to define the origin. All non-H bond distances and angles were subjected to restraints, based on a Mercury Mogul Geometry Check (Bruno et al., 2004; Sykes et al., 2011). The Mogul average and standard deviation for each quantity were used as the restraint parameters. The restraints contributed 5.5% to the overall χ^2 . The hydrogen atoms were included in calculated positions, which were recalculated during the refinement using Materials Studio (Dassault Systèmes, 2023). The U_{iso} of the heavy atoms were grouped by chemical similarity. The U_{iso} for the H atoms were fixed at $1.3 \times$ the U_{iso} of the heavy atoms to which they are attached. The peak profiles were described using a uniaxial microstrain model, with [010] as the unique axis. The background was modeled using a 6-term shifted Chebyshev polynomial, with peaks at 3.02 and 10.20° to model the narrow and broad scattering from the Kapton capillary and any amorphous component.

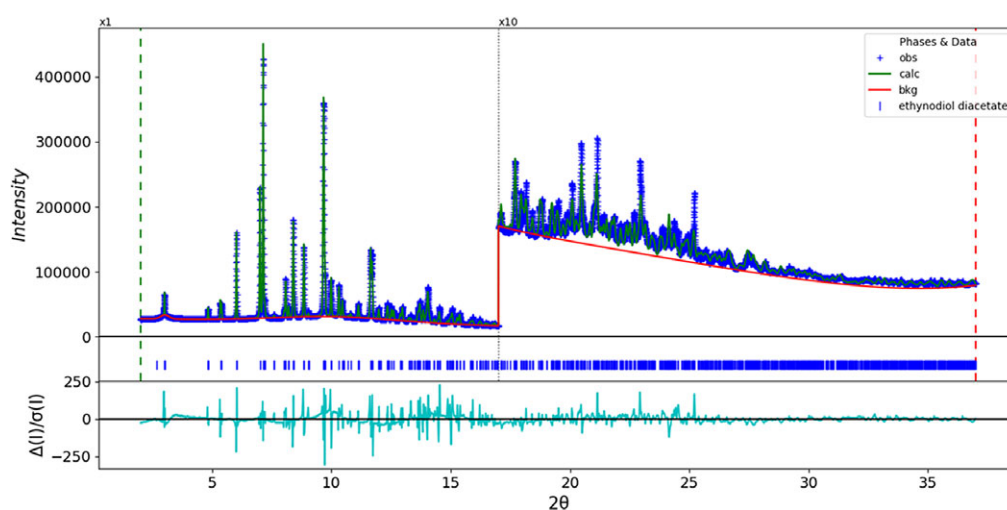


Figure 2. The Rietveld plot for ethynodiol diacetate. The blue crosses represent the observed data points, and the green line is the calculated pattern. The cyan curve is the normalized error plot, and the red line is the background curve. The blue tick marks indicate the ethynodiol diacetate peak positions. The vertical scale has been multiplied by a factor of $10 \times$ for $2\theta > 17.0$.

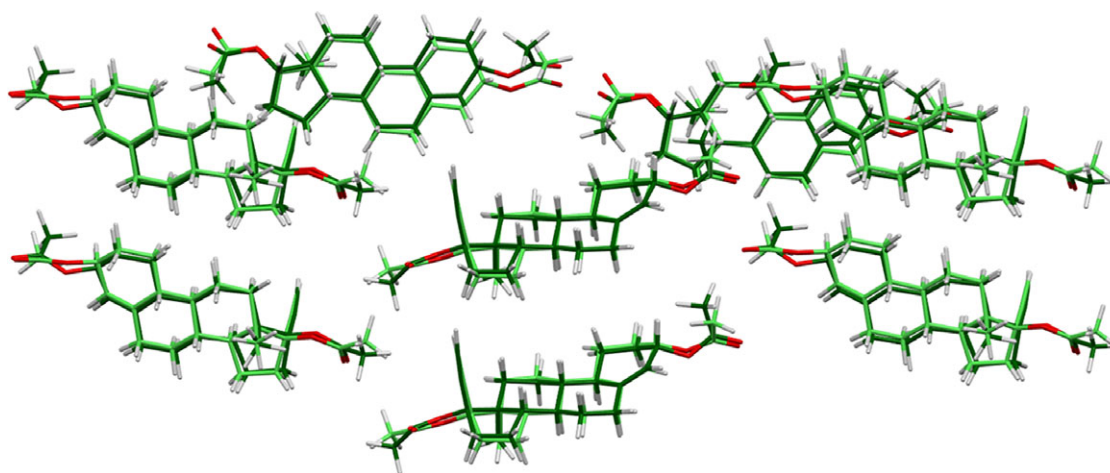


Figure 3. Comparison of the Rietveld-refined (colored by atom type) and VASP-optimized (light green) structures of ethynodiol diacetate, calculated using the Mercury CSD-Materials/Search/Crystal Packing Similarity tool. The root-mean-square Cartesian displacement is 0.462 \AA . Image generated using Mercury (Macrae et al., 2020).

The final refinement of 197 variables using 14,001 observations and 158 restraints yielded the residual $R_{wp} = 0.05683$. The largest peak (1.93 Å from C82) and hole (2.01 Å from C71) in the difference Fourier map were 0.29(8) and $-0.30(8) e\text{\AA}^{-3}$, respectively. The final Rietveld plot is shown in Figure 2. The largest features in the normalized error plot are in the asymmetries, shapes, and positions of some of the strong low-angle peaks. These misfits probably indicate subtle changes in the specimen during the measurement.

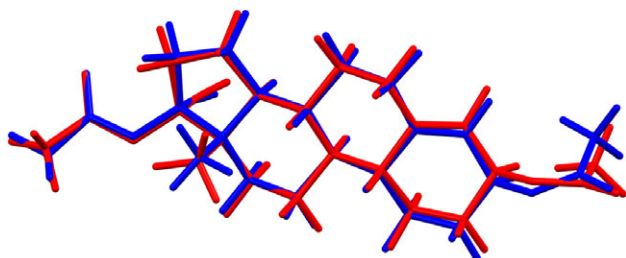


Figure 4. Comparison of the Rietveld-refined (red) and VASP-optimized (blue) structures of molecule 1 in ethynodiol diacetate. The root-mean-square Cartesian displacement is 0.261 Å. Image generated using Mercury (Macrae et al., 2020).

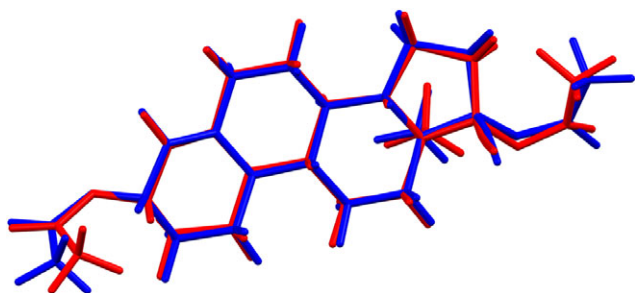


Figure 5. Comparison of the Rietveld-refined (red) and VASP-optimized (blue) structures of molecule 2 in ethynodiol diacetate. The root-mean-square Cartesian displacement is 0.309 Å. Image generated using Mercury (Macrae et al., 2020).

The crystal structure of ethynodiol diacetate was optimized (fixed experimental unit cell) with density functional theory techniques using Vienna ab initio simulation package (VASP) (Kresse and Furthmüller, 1996) through the MedeA graphical interface (Materials Design, 2024). The calculation was carried out on 32 cores of a 144-core (768 GB memory) HPE Superdome Flex 280 Linux server at North Central College. The calculation used the generalized gradient approximation with the Perdew–Burke–Ernzerhof functional (GGA-PBE) functional, a plane-wave cutoff energy of 400.0 eV and a k -point spacing of 0.5 \AA^{-1} leading to a $1 \times 2 \times 1$ mesh and took ~ 32.2 hours. Single-point density functional calculations (fixed experimental cell) and population analysis were carried out using CRYSTAL23 (Erba et al., 2023). The basis sets for the H, C, and O atoms in the calculation were those of Gatti et al. (1994). The calculations were run on a 3.5-GHz PC using 8 k -points and the Becke, three-parameter, Lee–Yang–Parr (B3LYP) functional and took ~ 7.3 hr.

III. RESULTS AND DISCUSSION

There are two independent ethynodiol diacetate molecules in the asymmetric unit. The root-mean-square (rms) difference of the non-H atoms in the Rietveld-refined and VASP-optimized structures was calculated using the Mercury CSD-Materials/Search/Crystal Packing Similarity tool is 0.462 Å (Figure 3). The rms Cartesian displacements of the non-H atoms in the Rietveld-refined and VASP-optimized

TABLE I. Unusual bond angles (Mogul) in ethynodiol diacetate

Angle	Value	Average	Z-score
C23–O2–C27	123.7	117.7(14)	4.1
O2–C27–C28	120.4	110.1(18)	5.2
C69–O61–C84	125.5	119.2(14)	4.5
C83–O62–C87	122.6	117.7(14)	3.3
O61–C84–O63	116.8	123.6(19)	3.5
O61–C84–C86	120.2	110.7(20)	4.8
O62–C87–C88	119.2	111.0(18)	4.6

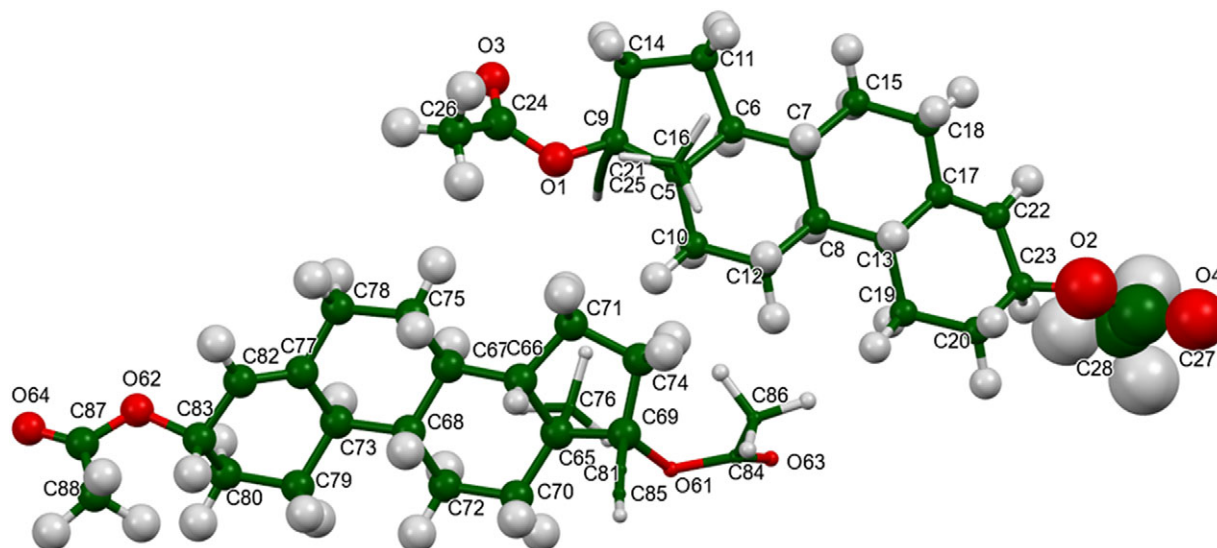


Figure 6. The asymmetric unit of ethynodiol diacetate, with the atom numbering. The atoms are represented by 50% probability spheroids. Image generated using Mercury (Macrae et al., 2020).

structures of molecule 1 and molecule 2, calculated using the Mercury Calculate/Molecule Overlay tool, are 0.261 and 0.309 Å (Figures 4 and 5). The largest differences are in the acetate side chains at the periphery of the molecules. The agreements are at the upper end of the normal range for correct structures (van de Streek and Neumann, 2014). The asymmetric unit is illustrated in Figure 6. The U_{iso} of the acetate group O2/O4/C27/C28 in molecule 1 are larger than those of the other atoms and may indicate a disorder of this side chain. Since we need an ordered model to carry out the DFT optimization, we chose to refine an ordered model, rather than attempt to model any disorder. The remaining discussion will emphasize the VASP-optimized structure.

All of the bond distances and most of the bond angles and torsion angles fall within the normal ranges indicated by a Mercury Mogul Geometry check (Macrae et al., 2020). The unusual bond angles are reported in Table I. All of the unusual angles occur in the acetate side chains. Torsion angles involving rotation about the C9–C21 and C69–C81 bonds have few/no hits, the result of the cyano groups. Torsion angles involving rotation about the O2–C27, O61–C84, and O62–C87 bonds are flagged as unusual and occur on the tails of

0/180° peaks in the distributions of similar torsion angles. These occur in the acetate side chains. The acetate groups seem to be unusual.

The two molecules have different conformations (Figure 7). The rms difference is 0.782 Å, and the major differences are in the conformations of the acetate groups. Quantum chemical geometry optimizations of isolated ethynodiols diacetate molecules (DFT/B3LYP/6-31G*/water) using Spartan'24 (Wavefunction, Inc., 2023) indicated that the two molecules are essentially identical in energy and converge to the same local minimum. The global minimum-energy conformation (Merck molecular force field [MMFF] force field) is 10.2 kcal/mol lower in energy and has a more linear orientation of the acetate group on the 6 ring. Even though weak, intermolecular interactions apparently affect the solid-state conformations.

The crystal structure (Figures 8 and 9) consists of alternating layers of molecule 1 and molecule 2 parallel to the (101) plane. The mean planes of molecules 1 and 2 are approximately (6, −7, 16) and (6, −2, 7), respectively. Analysis of the contributions to the total crystal energy of the structure using the Forcite module of Materials Studio (Dassault Systèmes, 2023) indicates that the intramolecular energy is dominated by angle distortion terms (as expected for a fused ring system), but that bond and torsion terms are also significant. The intermolecular energy is dominated by electrostatic attractions.

There are no classical hydrogen bonds in the structure (Table II). Only a small number of C–H...O and C–H...C (involving the cyano groups) hydrogen bonds represent directional interactions among the molecules. Fingerprint plots (Spackman et al., 2021) indicate that H...H contacts make up 66.66% of the intermolecular contacts (Figure 10) and that O...H contacts make up 13.4% of the interactions. The molecules do not interact strongly with each other.

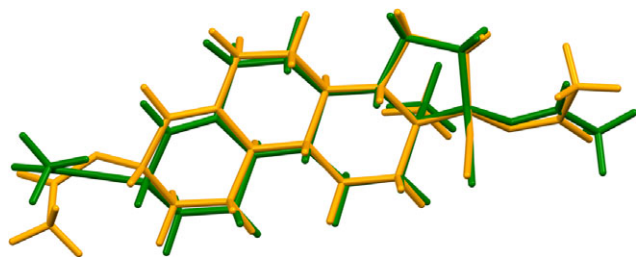


Figure 7. Comparison of molecule 1 (green) and molecule 2 (orange) of ethynodiols diacetate. Image generated using Mercury (Macrae et al., 2020).

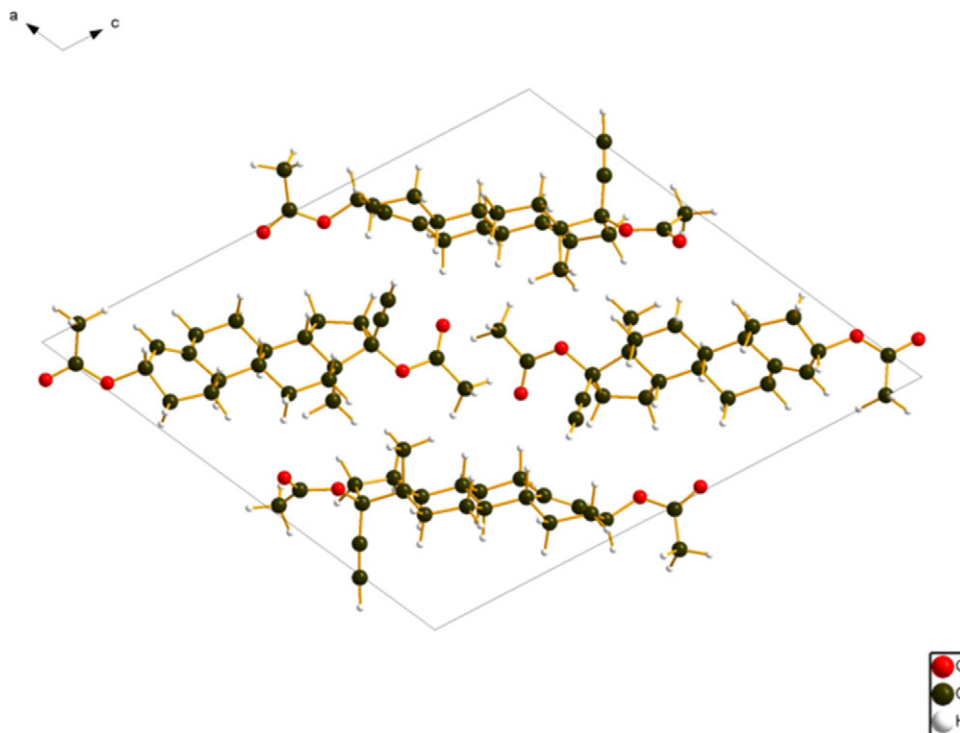


Figure 8. The crystal structure of ethynodiols diacetate viewed down the b -axis. Image generated using Diamond (Crystal Impact, 2023).

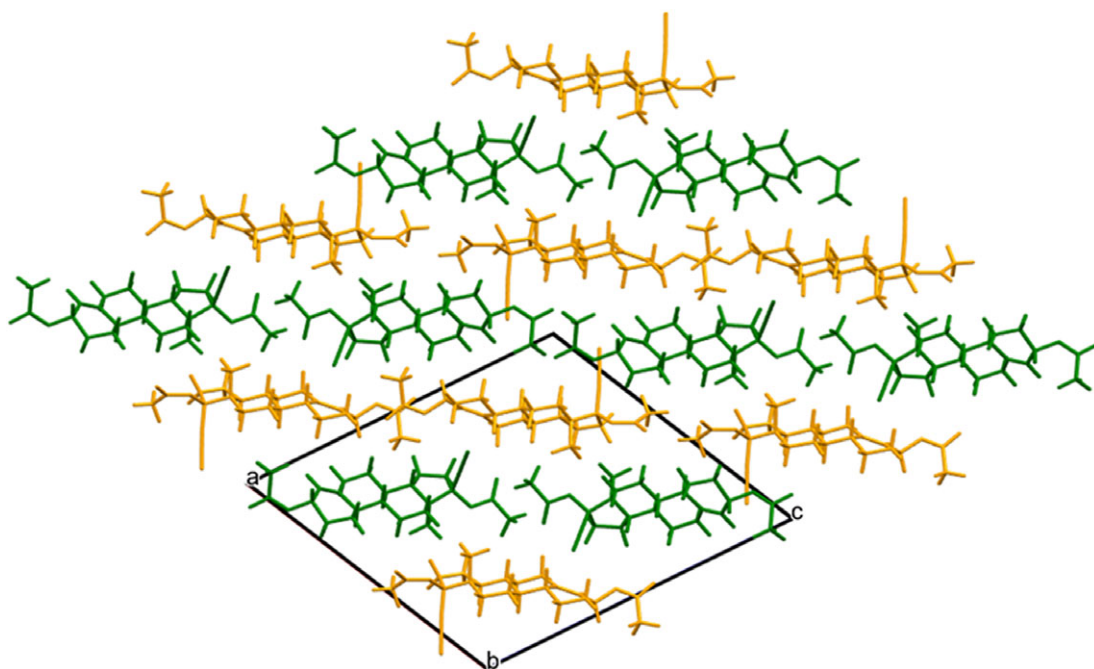


Figure 9. The crystal structure of ethynodiols diacetate, viewed down the b -axis. Molecule 1 is in green, and molecule 2 is in orange. Image generated using Mercury (Macrae et al., 2020).

TABLE II. Hydrogen bonds (CRYSTAL23) in ethynodiols diacetate

H-bond	D–H, Å	H...A, Å	D...A, Å	D–H...A, °	Mulliken overlap, e
C6–H29...C21	1.108	2.603 ^a	3.003	100.1	0.010
C10–H33...C21	1.102	2.490 ^a	2.918	101.6	0.010
C10–H33...C25	1.102	2.787 ^a	3.557	126.6	0.010
C26–H55...O3	1.102	2.788	3.842	159.9	0.010
C82–H112...O64	1.095	2.809	3.732	141.8	0.012

^aIntramolecular.

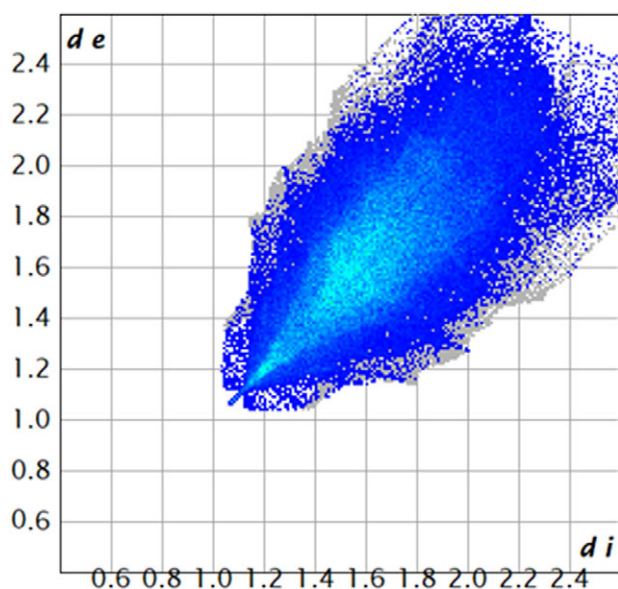


Figure 10. A fingerprint plot for ethynodiols diacetate, illustrating the H...H contacts. Image generated using CrystalExplorer (Spackman et al., 2021). d_e is the distance from the Hirshfeld surface to the nearest nucleus outside the surface. d_i is the corresponding distance to the nearest nucleus inside the surface. The blue and cyan points represent H...H contacts, and the total area of these contacts is 66.66% of the area of the Hirshfeld surface.

The volume enclosed by the Hirshfeld surface of ethynodiols diacetate (Figure 11; Hirshfeld, 1977; Spackman et al., 2021) is 1098.90 Å³, 98.98% of 1/2 of the unit cell volume. The packing density is lower than normal, reflected by the low density of 1.150 g/cm³. There are very few close contacts (red in Figure 11). The volume/non-hydrogen atom is larger than normal at 19.8 Å³.

The Bravais–Friedel–Donnay–Harker algorithm (Bravais, 1866; Friedel, 1907; Donnay and Harker, 1937) suggests that we

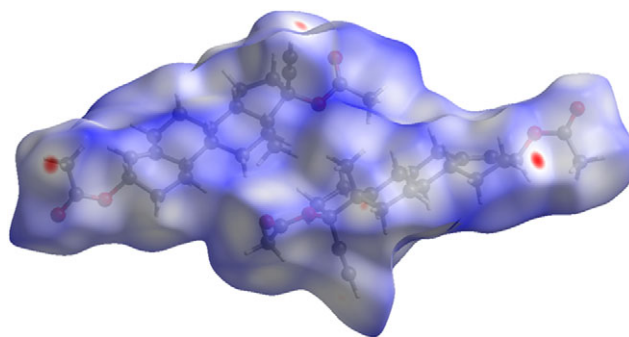


Figure 11. The Hirshfeld surface of ethynodiols diacetate. Intermolecular contacts longer than the sums of the van der Waals radii are colored blue, and contacts shorter than the sums of the radii are colored red. Contacts equal to the sums of radii are white. Image generated using CrystalExplorer (Spackman et al., 2021).

might expect elongated morphology for ethynodiol diacetate, with [010] as the major axis. A second-order spherical harmonic model was included in the refinement. The texture index was 1.004(0), indicating that the preferred orientation was insignificant in this rotated capillary specimen.

DEPOSITED DATA

The powder pattern of ethynodiol diacetate from this synchrotron data set has been submitted to the International Centre for Diffraction Data (ICDD) for inclusion in the Powder Diffraction File. The Crystallographic Information Framework (CIF) files containing the results of the Rietveld refinement (including the raw data) and the density functional theory (DFT) geometry optimization were deposited with the ICDD. The data can be requested at pdj@icdd.com.

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CONFLICTS OF INTEREST

The authors have no conflicts of interest to declare.

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